For the Use of a Registered Medical Practitioner only

PRESCRIBING INFORMATION

Gestid_{Plus} Suspension

(Calcium Carbonate, Sodium Bicarbonate, Dried Aluminium Hydroxide, Magnesium Hydroxide and Simethicone Suspension)

COMPOSITION

Gestid Plus Suspension (Lemon Mint)

Each 5 ml contains:	
Calcium Carbonate BP	125 mg
Sodium Bicarbonate BP	
Dried Aluminum hydroxide BP	
Magnesium Hydroxide BP	
Simethicone USP	20 mg
Saccharin Sodium BP	5 mg
Colours: Quinoline Yellow	
Colours. Quinoline reliow	

Contains Alginic acid. flavours in a sugar-free base g.s.

Gestid Plus Suspension (Ginger aniseed mint)

Each 3 iii con	laiii5.		
Calcium Carbo	nate BP	125 mg	
Sodium Bicarb	onate BP	125 mg	
Dried Aluminur	n hydroxide BP	100 mg	
Magnesium Hy	droxide BP	100 mg	
Simethicone U	SP	20 mg	
Saccharin Sodi			
Colours: Quino	line Yellow & Br	rilliant Blue	
Contains Algini	c acid, flavours	in a sugar-free base of	1.S.

DESCRIPTION1,2,3,4,5,6

Gestid Plus Suspension contains the antacids calcium carbonate, sodium bicarbonate, dried aluminum hydroxide, magnesium hydroxide, the antiflatulent agent/surfactant simethicone, and

Calcium carbonate: The molecular formula is CaCO₃, with a molecular

Sodium Bicarbonate: It is chemically sodium hydrogen carbonate. The molecular formula is NaHCO₃ with a molecular weight of 84.006610

Dried aluminium hydroxide: It is chemically aluminum trihydrate. The molecular formula is AIH $_6$ O $_3$ with a molecular weight of 81.027378.

Magnesium Hydroxide: It has the molecular formula $Mg(OH)_2$ and

Simethicone: Simethicone is a poly (dimethylsiloxane) which is a polymer of 200-350 units of dimethylsiloxane, along with added silica gel.

Saccharin sodium: It is a flavoring agent and non-nutritive

INDICATIONS 10, 12

molecular weight 58.32.

Gestid Plus Suspension is indicated for the relief from indigestion.

DOSE AND METHOD OF ADMINISTRATION 10, 12, 13

Adults and children over 12 years:

10 ml to 20 ml after meals and at bedtime or as required.

An appropriate proportion of the adult dose should be given only on medical advice

Elderly:
There is no need for dosage reduction in the elderly.

USE IN SPECIAL POPULATIONS 8,10

Pregnancy
Safety of Gestid Plus Suspension in pregnancy has not been established. Caution should be exercised when prescribed to pregnant

Lactation

There is no data of **Gestid Plus Suspension** in lactation, should be given

CONTRAINDICATIONS^{10,1}

Gestid Plus Suspension is contraindicated in:

- Patients with hypersensitivity to any of the ingredients
- Patients with kidney failure

WARNINGS AND PRECAUTIONS, 9, 10, 12

Dried aluminium hydroxide and magnesium hydroxide: In patients with renal impairment, plasma levels of both aluminium and magnesium increase. In these patients, a long term exposure to high

doses of aluminium and magnesium salts may lead to osteomalacia or encephalopathy with seizures and dementia, bone disease, and microcytic anaemia.

Aluminium hydroxide may be unsafe in patients with porphyria undergoing hemodialysis.

There is little evidence that aluminium containing antacids are a risk factor $\,$

Hypermagnesaemia may occur, usually in patients with renal impairment

Sodium Bicarbonate

Bicarbonate should not generally be given to patients with metabolic or respiratory alkalosis, hypocalcaemia, or hypochlorhydria. During treatment of acidosis, frequent monitoring of serum-electrolyte concentrations and acid-base status is

Sodium-containing salts should be given extremely cautiously to patients with heart failure, oedema, renal impairment, hypertension, eclampsia, or aldosteronism

Care needs to taken in treating patients with hypercalcaemia, nephrocalcinosis, and recurrent calcium containing renal calculi.

Gestid Plus Suspension contains sorbitol. Patients with rare hereditary problems of fructose intolerance should not take this

DRUG INTERACTIONS .8, 10, 11, 12

Aluminium compounds used as antacids interact with many other drugs, both by alterations in gastric pH and emptying, and by direct adsorption and formation of complexes that are not absorbed. Interactions can be inimised by giving the aluminium compound and any other medication 2

This product may interfere with the absorption of tetracyclines, iron salt. chloroquine, penicillamine, phenothiazines and quinolone antibacterials when given concomitantly.

Black Gestid Plus Susp PIL PIL size - 140 x 200 mm Market: Nigeria **SPIL/PKGDEV - CG17/Jul/2018-V01, V02** The absorption of aluminium from the gastrointestinal tract may be enhanced if aluminium compounds are taken with citrates or ascorbic acid especially in patients with renal impairmen

Omeprazole significantly reduces the absorption of calcium from a calcium carbonate supplement given on an empty stomach

Thiazide diuretics increase the risk of the milk-alkali syndrome in patients taking moderately large doses of calcium carbonate.

Sodium bicarbonate enhances lithium excretion. Concurrent use is not

Alkalinisation of the urine leads to increased renal clearance of acidic drugs such as salicylates, tetracyclines, and barbiturates.

UNDESIRABLE EFFECTS, 10,12

Patients sensitive to any of the ingredients may develop allergic manifestations such as urticaria or bronchospasm, anaphylactic or anaphylactoid reactions.

It is an astringent and may cause constipation. Large doses can cause intestinal obstruction. Excessive doses, or even normal doses in patients with low phosphate diets, may lead to phosphate depletion accompanied by increased bone resorption and hypercalciuria with the risk of

Magnesium hydroxide:

There have been reports of hypermagnesaemia and diarrhoea.

Calcium carbonate:

Calcium carbonate may occasionally cause constipation. Flatulence from released carbon dioxide may occur in some patients. Ingestion of large quantities of calcium carbonate may cause alkalosis, hypercalcaemia acid rebound, milk alkali syndrome or constipation. These usually occur

following larger than recommended dosages.

High doses or prolonged use may lead to gastric hypersecretion and acid rebound. Calcium carbonate can cause hypercalcaemia, particularly in patients with renal impairment or after high doses. Alkalosis may also occur as a result of the carbonate anion. There have been rare reports of the milk-alkali syndrome, and tissue calcification

Sodium bicarbonate

Excessive use of sodium bicarbonate at higher doses than recommended may lead to electrolyte imbalance, hypernatraemia, hypokalaemia and metabolic alkalosis, especially in patients with impaired renal function. Symptoms include mood changes, tiredness, slow breathing, muscle weakness, and irregular heartbeat. Muscle hypertonicity, twitching, and weathess, ain regular leadured. Musel in Protectionary, winding, and tetany may develop, especially in hypocalcaemic patients. Excessive doses of sodium salts may also lead to sodium overloading and hyperosmolality. Sodium bicarbonate given orally can cause stomach cramps, belching, and flatulence.

Simethicone: Gastrointestinal side effects are uncommon.

OVERDOSE. 10,12

Serious symptoms are unlikely following overdosage.

Symptoms of overdosage include nausea, vomiting, gastrointestinal irritation, diarrhoea/constipation. Treatment should be symptomatic and supportive.

PHARMACODYNAMIC AND PHARMACOKINETIC PROPERTIES.12

· Pharmacodynamics

Dried aluminium hydroxide and magnesium hydroxide are used as antacids. Aluminium hydroxide has constipating effects and magnesium hydroxide has laxative effects. These are given in combination to counteract their effects.

Calcium carbonate is an antacid. It rapidly reacts with gastric acid to

Sodium bicarbonate is a rapid onset, short acting antacid which neutralizes acid secretions in the gastrointestinal tract by reacting with hydrochloric acid to produce sodium chloride. During neutralization carbon dioxide is released, facilitating erucation which provides a sense Simethicone is used for the relief of flatulence and abdominal discomfort.

Aluminium hydroxide, given orally slowly reacts with the hydrochloric acid in the stomach to form soluble aluminium chloride, some of which is absorbed. The presence of food or other factors that decrease dastric emptying prolongs the availability of aluminium hydroxide to react and may increase the amount of aluminium chloride formed. Absorbed aluminium is eliminated in the urine and patients with renal failure are therefore at particular risk of accumulation. The aluminium compounds remaining in the gastrintestinal tract, which account for most of the dose form insoluble poorly, absorbed salts, which are excreted in the faeces. Magnesium hydroxide, given by mouth, reacts relatively rapidly with hydrochloric acid in the stomach to form magnesium chloride and water. About 30% of the magnesium ions are absorbed from the small intestine. calcium carbonate is converted to calcium chloride by gastric acid. Some of the calcium is absorbed from the intestines but about 85% is reconverted to insoluble calcium salts, such as the carbonate and is excreted in the faeces.

Administration of sodium bicarbonate by mouth causes neutralization of gastric acid with the production of carbon dioxide. Bicarbonate not involved in that reaction is absorbed and in the absence of a deficit of bicarbonate in the plasma, bicarbonate ions re excreted in the urine that is rendered alkaline with an accompanying diuresis.

Simethicone is not absorbed from the gastrointestinal tract.

STORAGE: Store below 30°C, away from direct sunlight. Do not freeze.

Keep all medicines out of reach of children

SUPPLY: Bottle pack of 100 ml and 180 ml

- Calcium carbonate: PubChem Compound Summary. Downloaded from
- Sodium Bicarbonate: PubChem Compound Summary. Downloaded from http://pubchem.ncbi.nlm.nih.gov/summary/summary.cgi?cid=516892&loc
- ium hydroxide: PubChem Compound Summary. Downloaded from Magnesium hydroxide: PubChem Compound Summary. Downloaded from http://pubchem.ncbi.nlm.nih.gov/summary/summary.cgi?cid=14791&loc=e
- Simethicone: PubChem Compound Summary Downloaded from ttp://pubchem.ncbi.nlm.nih.gov/summary/sumary.cgi?cid=6433516&loc=e
- Saccharin sodium: PubChem Compound Summary. Downloaded from ttp://pubchem.ncbi.nlm.nih.gov/summary/summary.cgi?cid=656582&loc=

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- tablets, orest Laboratories UK Limited, March 2010.
- UK Summary of product characteristics of Gaviscone double action liquidsReckitt Benckiser, UKJanuary 2006.

 AHFS Drug monograph: Simethicone: electronic version 2007. Information compiled in November, 2010.

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